

Plasma-surface coupled modelling of ammonia production in DC discharges



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Introduction

The search for a cleaner and more energy-efficient process for the **synthesis of ammonia (NH₃)** is a global concern, to which low-temperature plasma technology has emerged as a potential solution [1].

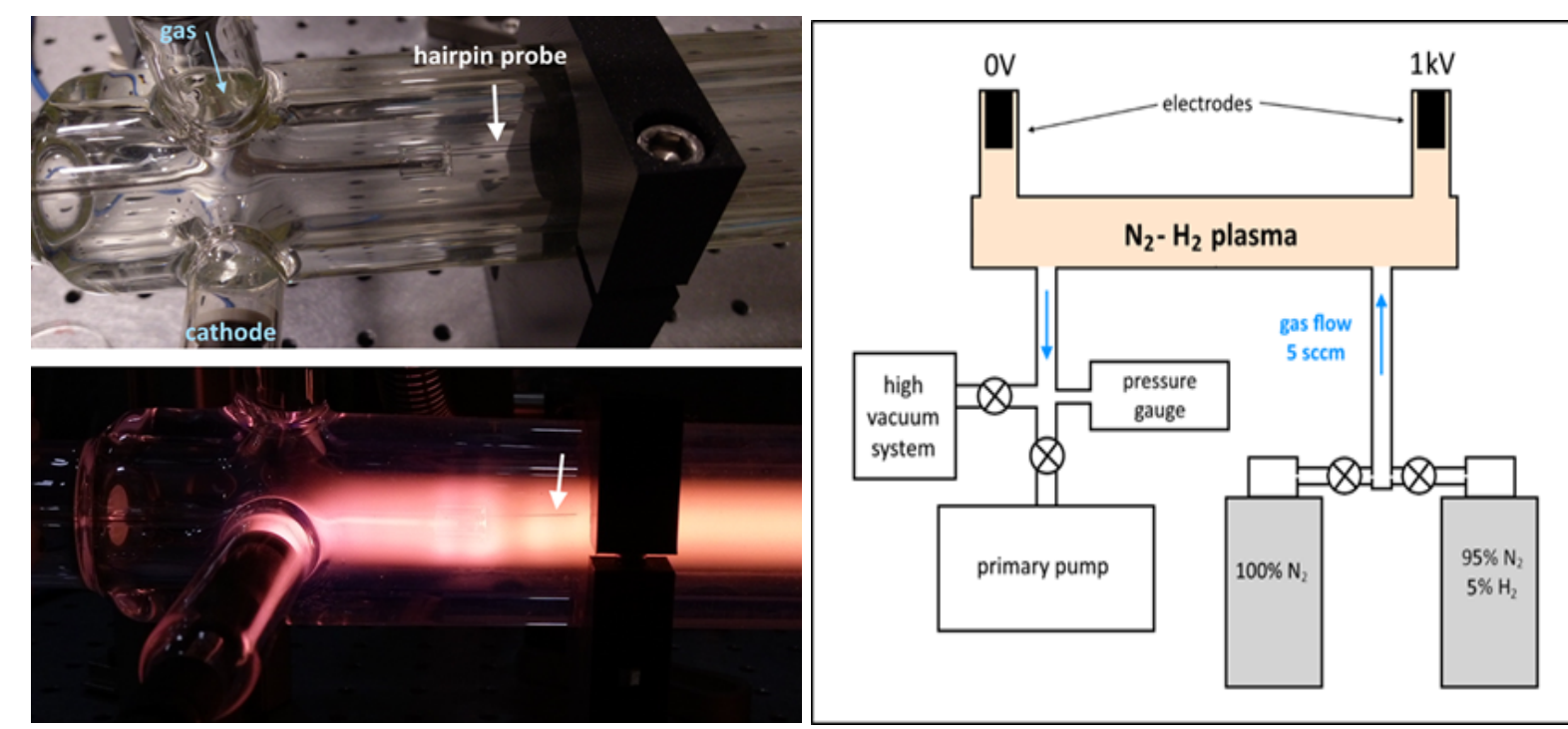
In this context, understanding the main kinetic paths leading to the plasma-assisted synthesis of NH₃ has topical interest, not only for the large-scale production of fertilizers at low cost, but also for the mitigation of ammonia generation in fusion machines.

It is currently believed that **plasma-surface interactions** could be the dominant mechanisms for this process [2]. In this work we address the kinetic modelling of N₂-H₂ plasmas, including the interplay between the volume and surface reactions.

We present the first steps of our research program on nitrogen-hydrogen plasmas, based on a **very complete kinetic scheme** for this system, which has started being revised and validated using complementary modelling and diagnostics analyses.

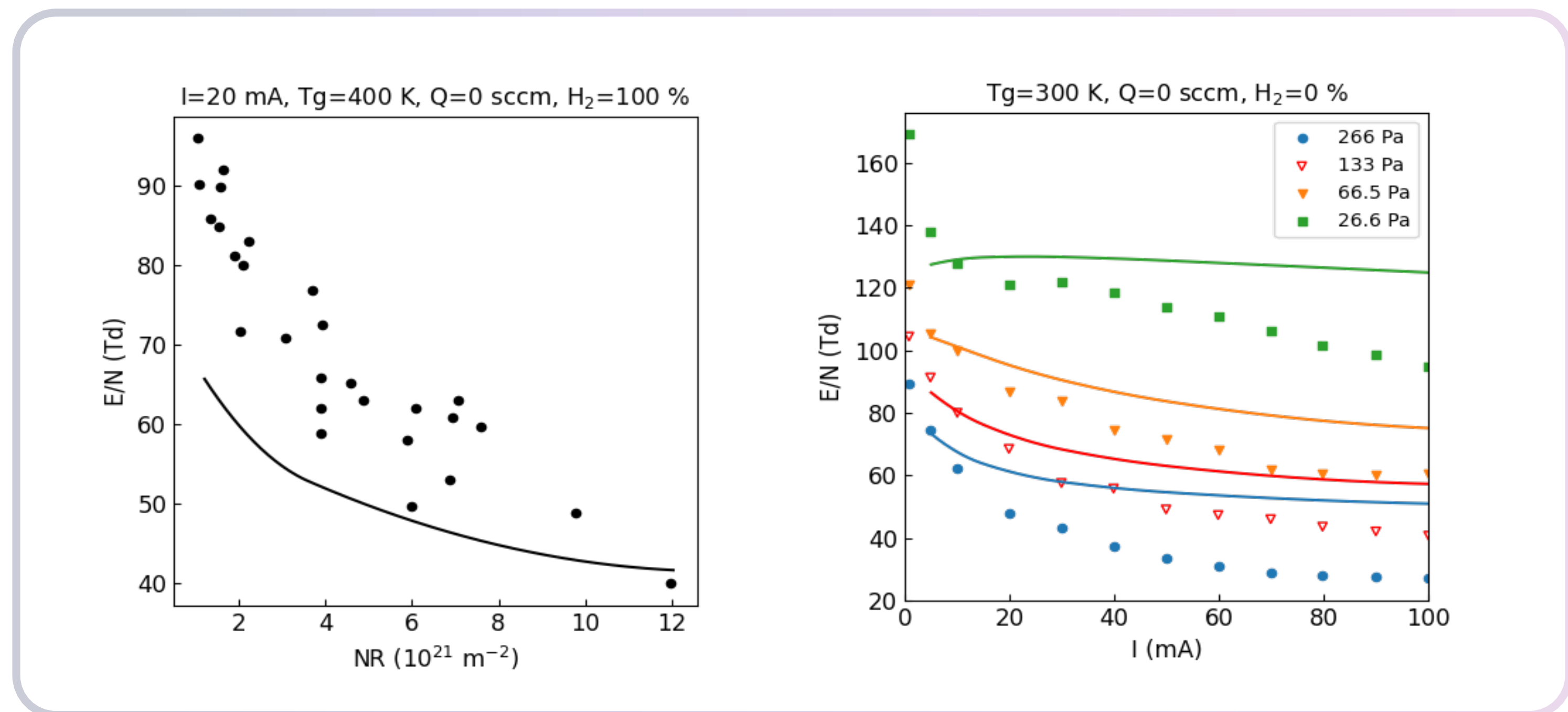
Experiments

We model **cylindrical DC glow discharges** ($L \sim 20$ cm long and $R \sim 1$ cm radius) with borosilicate glass walls [3,4], produced in N₂-H₂ gas mixtures at various H₂ concentrations (0-100%), gas flows ($Q = 0-600$ sccm), pressures ($p = 30-400$ Pa) and discharge currents ($I_{dc} = 5-100$ mA).

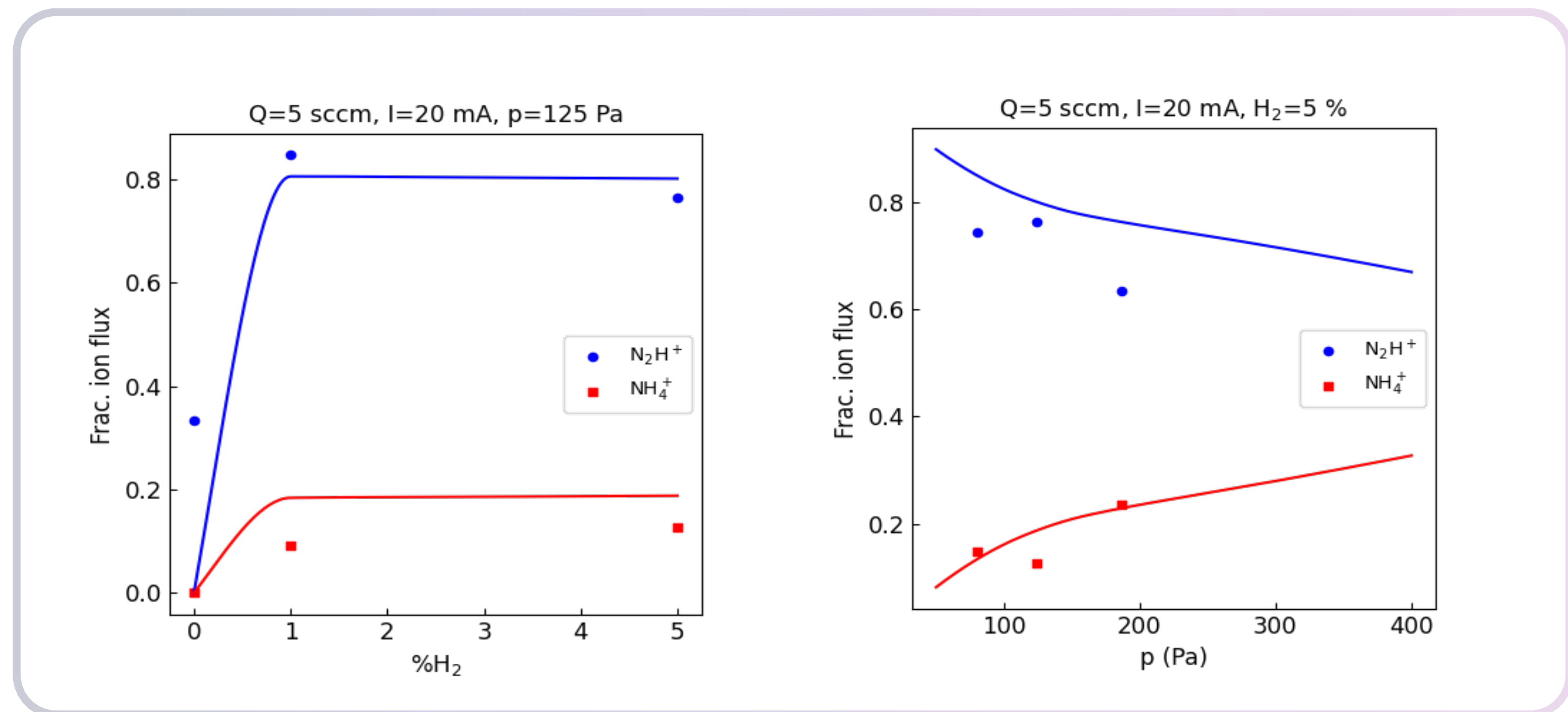


The figure shows the **experimental setup with the Laboratory of Plasma Physics (LPP)** in Paris. Typical measurements include: the **reduced electric field E/N** (where E is the electric field and N is the gas density), using the potential difference between two tungsten Langmuir probes immersed in the plasma; the **densities of ammonia (FTIR absorption)**, and of **H and N atoms (LIF measurements)**, and the **fractional fluxes of the main ion species**, using mass spectrometry.

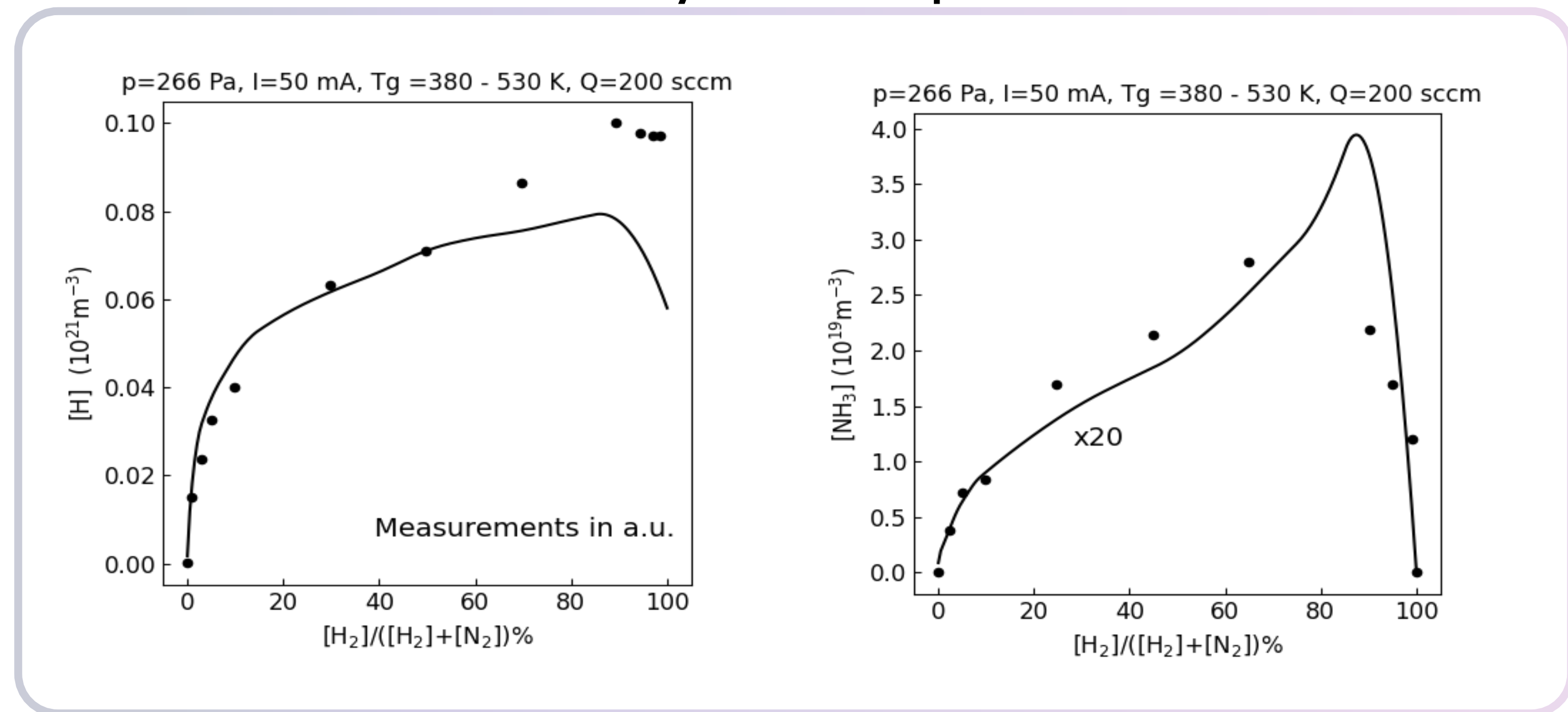
Plasma Characteristics



Fractional Ion Fluxes



Density of Neutral Species



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Simulations

Simulations use the **LisOn KInetics (LoKI) simulation tool** [5,7], developed under MATLAB® comprising

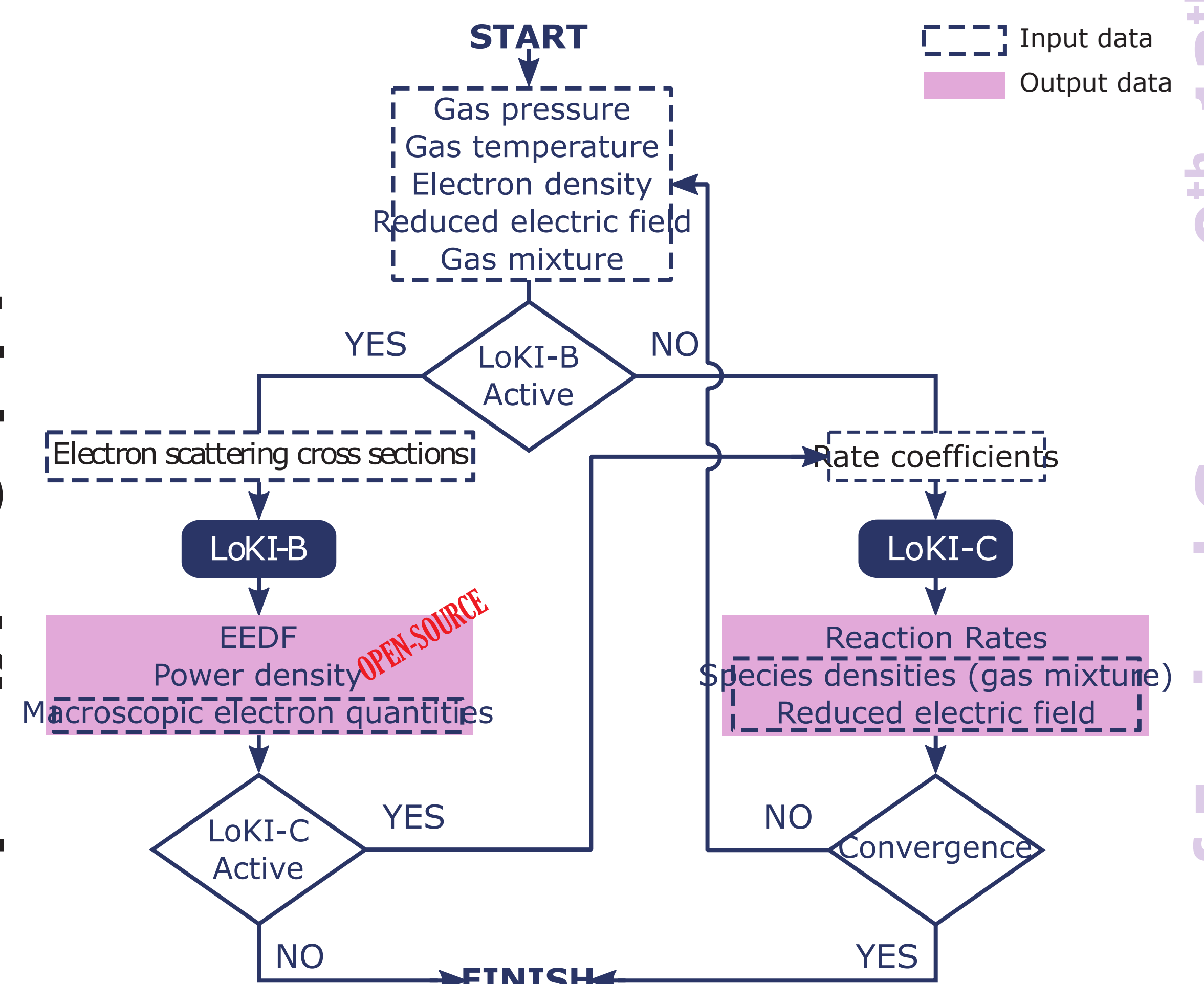
- **LoKI-B** [5,6] that solves a space independent form of the two-term **electron Boltzmann equation (EBE)** for non-magnetised non-equilibrium low-temperature plasmas excited by DC/HF, or time-dependent (non-oscillatory), electric fields, using electron-scattering cross sections that can be downloaded from the LXCat open-access website [8].

- **LoKI-C** [7] that solves the system of **zero-dimensional (volume average) rate balance equations** for the most relevant heavy species (charged and neutral) in the plasma, receiving as input data the kinetic schemes for the gas/plasma system under study, and using several modules to describe the mechanisms (collisional, radiative and transport) controlling the creation / destruction of species.

LoKI-B and LoKI-C are **coupled via a series of convergence cycles**, ensuring a self-consistent solution for the electron energy distribution function, the species densities, and the reduced electric field, for given pressure, mixture composition and discharge current.



LisOn Kinetics



The **kinetic scheme for nitrogen-hydrogen plasmas** [9], considers the main heavy species N₂(X, v=0-44), H₂(X, v=0-14), NH₃, N(⁴S), H(¹S), in addition to: 13 electronic excited states (6 for N₂, 2 for N and 5 for H); positive ions N⁺, N₂⁺, N₃⁺, N₄⁺, H⁺, H₂⁺, H₃⁺, N₂H⁺, NH⁺, NH₂⁺, NH₃⁺ and NH₄⁺; negative ions H⁻ and NH₂⁻; surface species H(S,F), N(S,F), NH(S), NH₂(S), (physically (F) or chemically (S) adsorbed on the wall); and other molecules and radicals.

The **surface kinetics is inspired by the mesoscopic model of Gordiets et al.** [4], considering physical adsorption/desorption, chemical adsorption, surface transport, and Eley-Rideal and Langmuir-Hinshelwood recombination processes. The surface density was assumed [S]+[F] = 10²⁰ m⁻².

Results and discussion

The adjacent figures present examples of comparisons between simulations and measurements, obtained for the working conditions mentioned above.

The **"Plasma Characteristics" panel** shows measurements and calculations of **E/N in pure H₂** (as a function of NR) and **pure N₂** (as a function of I_{dc}). Model predictions underestimate the values of E/N for pure H₂, whereas showing good agreement for pure N₂ at low discharge currents. An improvement in the results may require revising the kinetic description of the dominant ion species: N₄⁺ and N₂⁺ in pure nitrogen; H₃⁺ in pure hydrogen; N₂H⁺ and NH₄⁺ in the N₂-H₂ mixture (see **"Fractional Ion Fluxes" panel**). In particular, decreasing the production rate coefficient of H₃⁺, and increasing the production rate of N₂(A) and N₂(a') may increase / decrease the value of E/N for pure H₂ / N₂ cases.

Note that the model correctly predicts the **fractional ion fluxes of N₂H⁺ and NH₄⁺**, as a function of H₂ concentration (0-5%) and pressure (75-400 Pa), as shown in the **"Fractional Ion Fluxes" panel**.

The **"Density of Neutral Species" panel** presents results for the **H-atom (left) and the NH₃ (right) densities**, as a function of the H₂ concentration. The LIF measurements of [H] were given in arbitrary units, so they have been normalized to the calculated values at 50% H₂ for comparison purposes. In this case, simulations show good qualitative agreement with the experiment. The **absolute measurements of [NH₃]** employed photofragment translational spectroscopy associated with LIF of H(¹S), calibrated by VUV absorption spectroscopy on the Lyman- α (121nm) line [10]. In this case, model predictions underestimate the measurements by 20x, but once the simulated curve is multiplied by this factor it agrees very well with the experiment.

The **mesoscopic surface kinetic model** efficiently describes the ammonia production by plasma-surface interaction. The physical adsorption of N,H atoms on physical vacant sites F(v) is crucial to engage the surface pathway leading to the most important ammonia production channels: Eley-Rideal and Langmuir-Hinshelwood recombination. A revision of the model mesoscopic parameters may significantly improve the results.

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